III-62 On lime-stabilized fly ashes by X-ray microanalyzer in relation to strength development

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INTRODUCTION

An attempt was made on utilizing fly ash, which chemical compositions are shown in Table 1. Lime (L) and lime mixed with chemical additives (G) were used as stabilizers. The quest and SEM-EDX analysis were performed to clarify strength-imparting mechanism due to so-called Ettringite, Monosulphate, and C-S-H phases. The results show that microstructures changed as strength was developing, and that stabilized fly ash may be used as material for constructing subgrade, subbase, etc.

TABLE 1 Chemical compositions of fly ash used

oxide	SiO2	A1203	CaO	TiO2	K 2 O	Na ₂ 0	Fe 2 O 3	MgO	S03
(%)	49.3	22.0	5.40	1.23	1.20	2.01	4.40	1.10	0.70

SAMPLE CONDITIONS

The samples were made by adding stabilizers into fly ash; only lime-fly ash (FL) and lime added with Carbonated-Aluminium Salt - fly ash (FG). Mix ratio of L and G to fly ash was 20:80. Treated water content was 60% of total weight (see ref.1 for more details).

COMPOSITIONAL ANALYSIS BY X-RAY MICROANALYZER

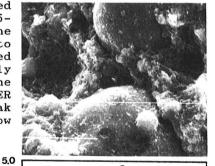
The failed sample obtained from q test was performed with a HORIBA X-ray microanalyzer. Peak element ratio (PER) and surface area ratio (SAR), defined as ratios of intensities of each constitutive element to Ca on specific section and entire viewing surface respectively, are proposed as two significant ratios.

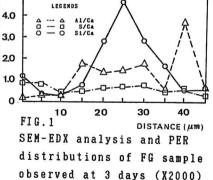
RESULTS AND DISCUSSION

It has been accepted that C-S-H product cured at ordinary tempt. has a Si/Ca ratio of 0.5-0.7. Based on the results obtained from line analysis, predicted ratios corresponding to dominant reaction products are then proposed in Table 2. PER of higher value (generally larger than 1.5) was found in unreacted zone or particle. A notch-like distribution of PER across a section (see Fig.1) can refer to weak section or weak seam, resulting in a low

TABLE 2 Predicted element ratios of reaction

REACTION PRODUCT	GUENZGAL BODWILLAD	ELEMENT RATIO			
REACTION PRODUCT	CHEMICAL FORMULAR	FROM FORMULAR	PREDICTED FROM EDX		
C-S-H groups	3Ca0 • 2Si02 • 3H20	Si/Ca = 0.66	Si/Ca = 0.50-0.70		
Ettringite and	3CaO·Al ₂ O ₃ ·3CaSO ₄ · 32H ₂ O	A1/Ca = 0.33 S/Ca = 0.50	A1/Ca = 0.20-0.30		
Monosulphate groups	3CaO·Al ₂ O ₃ ·CaSO ₄ · 12H ₂ O	Al/Ca = 0.50 S/Ca = 0.25	S/Ca = 0.10-0.50		
Calcium carboaluminate	3CaO·Al ₂ O ₃ ·CaCO ₃ · 12H ₂ O	A1/Ca = 0.25	Al/Ca = 0.25-0.35		
Calcium carbonate	CaCO ₈		Si/Ca < 0.50		
Calcium hydroxide	Ca (OH) 2		A1/Ca < 0.20		
Calcium sulphate	CaSO₄		S/Ca < 0.10		





large stability but deformation. However, it was apparent that PER tended to reduce its value into the ranges of PER (Table 2) as predicted periods increased (see Fig.2). SAR can also represent rigidity of microsection the value had a prone to fall predicted PER ranges (see Table 3). only lime -fly ash, neither the PER nor SAR tended to incline into the predicted ranges though curing time increased (see Fig. 3). The strength characteristic curves of two mixtures are given Fig. 4. It is apparent that fly ash mixed with G may even be used as base course material, and thus small chemical additives added into lime could improve hardening characteristic of

TABLE 3 SAR of FG sample at various curing

days	5					
ELEMENT RATIO	SURFACE AREA RATIO					
	3 DAYS	7 DAYS	28 DAYS	90 DAYS		
Al/Ca	0.37	0.22	0.24	0.24		
S/Ca	0.38	0.12	<0.1	0.18		
Si/Ca	0.75	0.51	0.25	0.54		

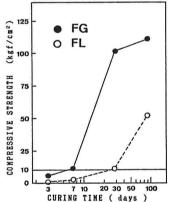
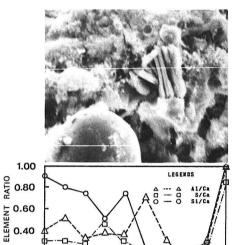


FIG.4 Strength-characteristic curves of stabilized fly ash

CONCLUSION

1) PER and SAR may be used as indicators for strength test for stabilized soil in the future. The test can be assumed to be a non-destructive testing method since it requires only a few specimen for analysis.

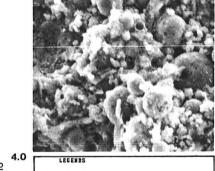


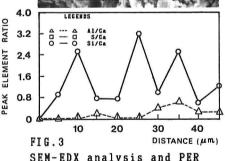
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FIG. 2

SEM-EDX analysis and PER

distributions of FG sample
observed at 90 days (X2000)





SEM-EDX analysis and PER distribution of FL sample observed at 90 days (X2000)

2) A knowledge of PER associated with SAR can elucidate reaction mechanism, including strength-imparting mechanisms of stabilized soil.

REFERENCE

1) KAMON, M. et al (1987); "The utilization of various waste materials by cement hardening", Review of The 41st General Meeting, Technical Session, CAJ, pp.84-87.